

## **SECTION 2**

# **Water Quality Monitoring Programs**

In 2001, the King County Department of Natural Resources and Parks (DNRP) operated three wastewater treatment plants and two CSO treatment plants with outfalls discharging directly into Puget Sound marine waters. The Clean Water Act states that all sewage treatment plants that discharge effluent from a point source into surface waters must have a National Pollutant Discharge Elimination System (NPDES) permit. In Washington, the Washington State Department of Ecology (DOE) administers the NPDES permit program by delegation from the U.S. Environmental Protection Agency (EPA).

King County's NPDES permits set limits on the quality of treated wastewater that is discharged through individual outfalls. In compliance with NPDES requirements and to verify that the facilities are meeting the goals of the Clean Water Act, King County has conducted an extensive point source monitoring program for over 20 years to assess the quality of each facility's effluent, the receiving water and sediments around each outfall, and nearby beaches.

Water quality may be affected by natural processes as well as by two types of pollution: point source and nonpoint source. Point source pollution is defined by its entry into the aquatic environment from a specific conveyance, such as an outfall pipe. Regulated pollution can be generated by a variety of industrial and municipal facilities, such as sewage treatment plants and manufacturing facilities. Nonpoint source pollution comes from any source that is not a point source and includes runoff or infiltration from streams, groundwater, storm water, etc. Land use, such as agricultural and urban usage, affects the quality of the runoff. King County's marine monitoring program is designed to assess potential effects from both types of pollution in both nearshore and offshore environments as well as assess ambient (background) conditions. The stations monitored by the marine program fall into one of two categories; ambient or point source. Within these categories, stations are classified as either beach (+3 m to -3 m mean lower low water), nearshore (-3 mean lower low water to 30 m), or offshore (bottom depth greater than 30 m).

Obtaining background data from areas in the receiving water that are not influenced by point sources is important in order to accurately evaluate the condition of the receiving water. King County has established an ambient monitoring program in the Central Puget Sound Basin, with stations well removed from the influence of King County treatment plant point source discharges. King County's goals for ambient monitoring are to better understand regional water quality and provide background data needed to identify trends that might indicate impacts from long-term cumulative pollution, or effects from King County outfalls.

Population growth in the greater Seattle area has necessitated planning for a new wastewater treatment plant to be located in northern King County or southern Snohomish County. This new treatment plant will include a marine outfall discharging secondary treated effluent to Puget Sound waters. Since 1998, King County has been conducting a marine outfall siting study (MOSS) as part of the planning process to assist with outfall siting and design and to assist with

evaluation of potential impacts to the marine environment. The MOSS sampling program consists of several components, including offshore water column profiling and nearshore/beach water quality monitoring. Sampling for this program started in December 1998 and will continue throughout 2002.

An overview of the County's point source, ambient, and MOSS monitoring programs for 2001 is provided in Table 2-1.

**Table 2-1. Summary of 2001 Marine Monitoring Programs**

Location	Matrix	Parameter	Number of Stations Sampled		
			Ambient	Point Source	MOSS
Beach	Water	Bacteria	15	5	12 <sup>1</sup>
		GWQP <sup>2</sup>	11	5	12 <sup>1</sup>
		Organics	--	--	3
		Metals	3	4	12 <sup>1</sup>
	Sediment	Organics	3	3	1 <sup>3</sup>
		Metals	3	3	1 <sup>3</sup>
		Conventionals <sup>4</sup>	3	3	1 <sup>3</sup>
	Shellfish	Organics	2	4	--
		Metals	2	4	--
		Bacteria	6	3	--
	Macroalgae	Metals	4	4	--
Nearshore	Water	Bacteria	2	1	--
		GWQP	2	1	--
Offshore	Water	Bacteria	4	5	10 <sup>5</sup>
		GWQP	4	5	10 <sup>5</sup>
	Sediment	Organics	--	19	--
		Metals	--	19	--
		Conventionals	--	19	--
		Benthic fauna	--	6	--

<sup>1</sup> Overlap with 3 ambient stations.

<sup>2</sup> GWQP = general water quality parameters. Includes temperature, salinity, nutrients, dissolved oxygen.

<sup>3</sup> Overlap with 1 ambient station.

<sup>4</sup> Conventionals include total solids, total volatile solids, total sulfide, total organic carbon, and grain size.

<sup>5</sup> Overlap with 3 ambient and 2 point source stations.

## 2.1 Point Source and Ambient Monitoring Programs

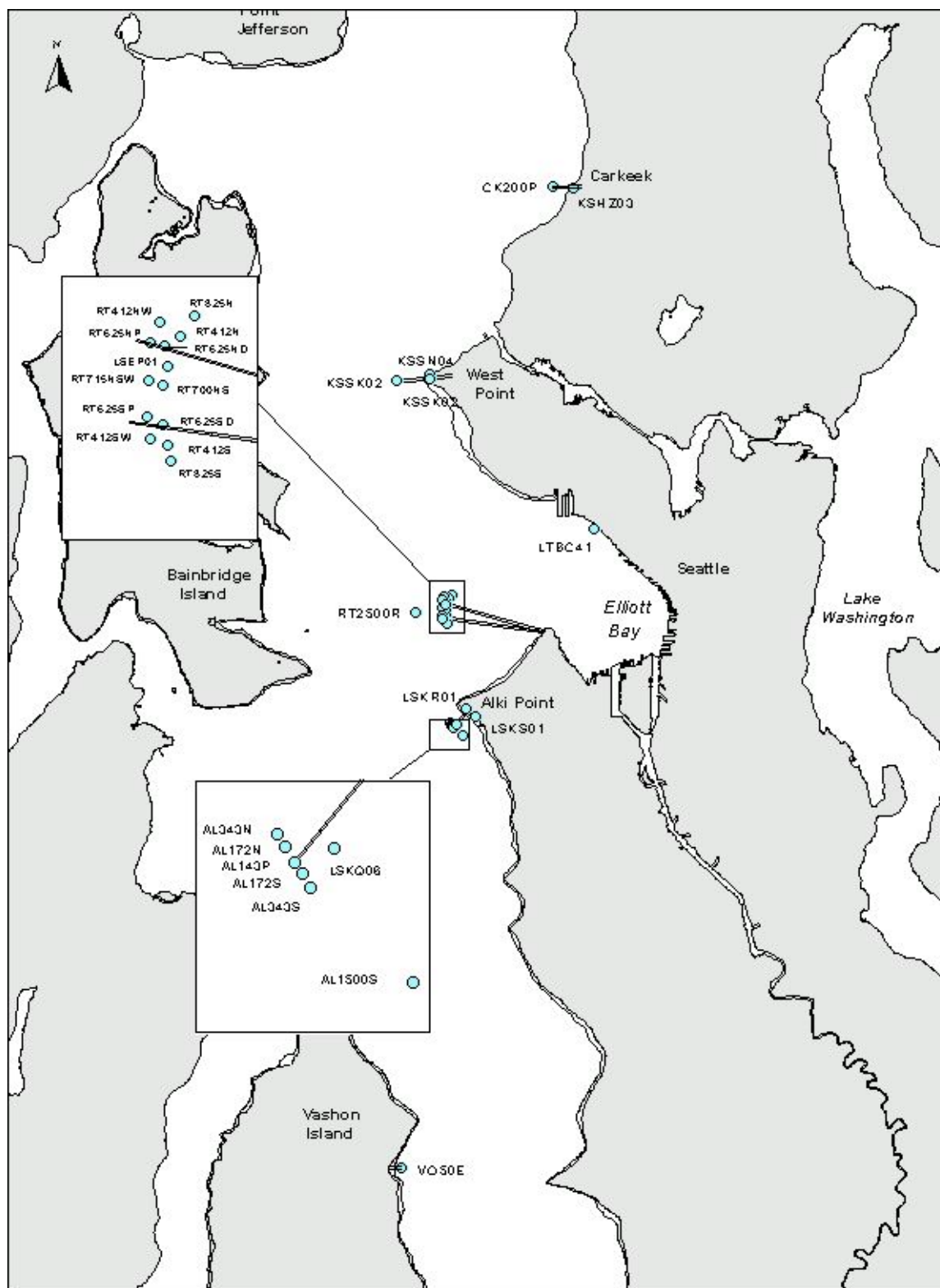
The ambient and point source monitoring programs focus on marine water and the underlying sediments. Many marine pollutants are in particulate form. As these contaminated particles settle out of the water column, pollutant concentrations in the underlying sediments tend to increase. Most sources of contamination are found in nearshore areas where pollutants tend to accumulate in sediments close to these sources. Benthic organisms that live on or in contaminated sediments tend to accumulate these contaminants through contact or ingestion (bioaccumulation). Pollutants also tend to concentrate as they move from one trophic level to the next (biomagnification), as contaminated organisms become prey to animals higher up in the food web. Contaminated sediments can have an important impact on human and marine environmental health, especially in nearshore areas which are generally high contact areas for marine organisms and people.

Nutrients and pathogens which may cause water quality problems in marine waters are also typically seen in nearshore areas in the vicinity of contamination sources. While excess nutrients do not cause immediate harm to organisms living in the water column, excess nutrients can increase the amount of phytoplankton and algae which can deplete oxygen when it decays to levels incapable of sustaining aquatic organisms.

### 2.1.1 Marine Point Source Monitoring Program

King County collected offshore and nearshore water samples and offshore sediment samples for the 2001 point source monitoring program. In addition, beach water, sediment, shellfish tissues, and macroalgae samples were collected in areas adjacent to the treatment plant outfalls. A total of 30 sites were sampled in 2001. Point source stations include those that are required by the County's NPDES permit (e.g., offshore sediment sites near the South Treatment Plant outfall) and those that are not required but are in close proximity to point source discharges (e.g., beach stations near the West Point Treatment Plant). Station locations are shown in Figure 2-1 and station coordinates are provided in Appendix F.

Water samples were analyzed monthly for temperature, salinity, clarity, dissolved oxygen, nutrients, chlorophyll, and bacteria. Offshore sediment samples were collected in late October to November and analyzed for organic compounds, metals, conventional parameters (total organic carbon, total solids, total volatile solids, ammonia-nitrogen, total sulfides, oil and grease, and grain size). Benthic fauna analysis was also conducted for several of the South TP outfall stations. Beach sediments were collected in August and analyzed for organic compounds, metals, conventional parameters (total organic carbon, total solids, total volatile solids, and grain size). Shellfish tissues were analyzed for organic compounds, metals, and bacteria. Macroalgae samples were analyzed for metals. Shellfish and algae samples were collected in August and additional shellfish samples for bacterial analyses were collected monthly from May to September. A summary of parameters measured and the frequency for each station is provided in Table 2-2.



**Figure 2-1. Point Source Monitoring Station Locations**

**Table 2-2. 2001 Point Source Stations, Parameters, and Frequency Measured**

STATION	LOCATION	OFFSHORE/ NEARSHORE/ BEACH	SEDIMENT				WATER				SHELLFISH			ALGAE
			Organics	Metals	Conventionals	Benthic comm.	Bacteria	GWQP *	Organics	Metals **	Organics	Metals	Bacteria	Metals
KSHZ03	Carkeek	Beach	◆ 1	◆ 1	◆ 1		◆ 12	◆ 12		◆ 1	◆ 1	◆ 1	◆ 5	◆ 1
CK200P	Carkeek	Offshore					◆ 12	◆ 12						
KSSN04	West Point	Beach	◆ 1	◆ 1	◆ 1		◆ 12	◆ 12		◆ 1	◆ 1	◆ 1	◆ 5	◆ 1
KSSN05	West Point	Beach					◆ 12	◆ 12		◆ 1	◆ 1	◆ 1		◆ 1
KSSK02	West Point outfall	Offshore					◆ 12	◆ 12						
LTBC41	Denny Way	Nearshore					◆ 12	◆ 12						
LSEP01	South Plant outfall	Offshore					◆ 12	◆ 12						
RT2500R	South Plant outfall	Offshore	◆ 1	◆ 1	◆ 1	◆								
RT412N	South Plant outfall	Offshore	◆ 1	◆ 1	◆ 1									
RT412NW	South Plant outfall	Offshore	◆ 1	◆ 1	◆ 1	◆								
RT412S	South Plant outfall	Offshore	◆ 1	◆ 1	◆ 1									
RT412SW	South Plant outfall	Offshore	◆ 1	◆ 1	◆ 1	◆								
RT625ND	South Plant outfall	Offshore	◆ 1	◆ 1	◆ 1									
RT625NP	South Plant outfall	Offshore	◆ 1	◆ 1	◆ 1	◆								
RT625SD	South Plant outfall	Offshore	◆ 1	◆ 1	◆ 1									
RT625SP	South Plant outfall	Offshore	◆ 1	◆ 1	◆ 1	◆								
RT700NS	South Plant outfall	Offshore	◆ 1	◆ 1	◆ 1									
RT715NSW	South Plant outfall	Offshore	◆ 1	◆ 1	◆ 1	◆								
RT825N	South Plant outfall	Offshore	◆ 1	◆ 1	◆ 1									
RT825S	South Plant outfall	Offshore	◆ 1	◆ 1	◆ 1									
LSKR01	Alki Point	Beach	◆ 1	◆ 1	◆ 1		◆ 12	◆ 12		◆ 1	◆ 1	◆ 1	◆ 5	◆ 1
LSKS01	Alki	Beach					◆ 12	◆ 12						
LSKQ06	Alki outfall	Offshore					◆ 12	◆ 12						
AL343N	Alki outfall	Offshore	◆ 1	◆ 1	◆ 1									
AL172N	Alki outfall	Offshore	◆ 1	◆ 1	◆ 1									
AL143P	Alki outfall	Offshore	◆ 1	◆ 1	◆ 1									
AL172S	Alki outfall	Offshore	◆ 1	◆ 1	◆ 1									
AL343S	Alki outfall	Offshore	◆ 1	◆ 1	◆ 1									
AL1500S	Alki outfall	Offshore	◆ 1	◆ 1	◆ 1									
VO50E	Vashon I. Outfall	Offshore					◆ 12	◆ 12						

\* GWQP = general water quality parameters. Includes nutrients, salinity, temperature, chlorophyll, dissolved oxygen, solids, transparency, photosynthetically active radiation for offshore/nearshore waters Includes nutrients, salinity, temperature for beach waters.

\*\* Dissolved metals

Both Point source and MOSS program stations

Both fecal coliform and enterococcus bacteria were measured in water and shellfish tissues. Enterococcus has been proposed as an indicator of bacterial contamination in marine waters, therefore, both types of bacteria were analyzed in anticipation of possible changes in bacterial monitoring procedures. Although there are no regulatory standards for fecal coliform bacteria in shellfish tissues, samples were analyzed for informative purposes.

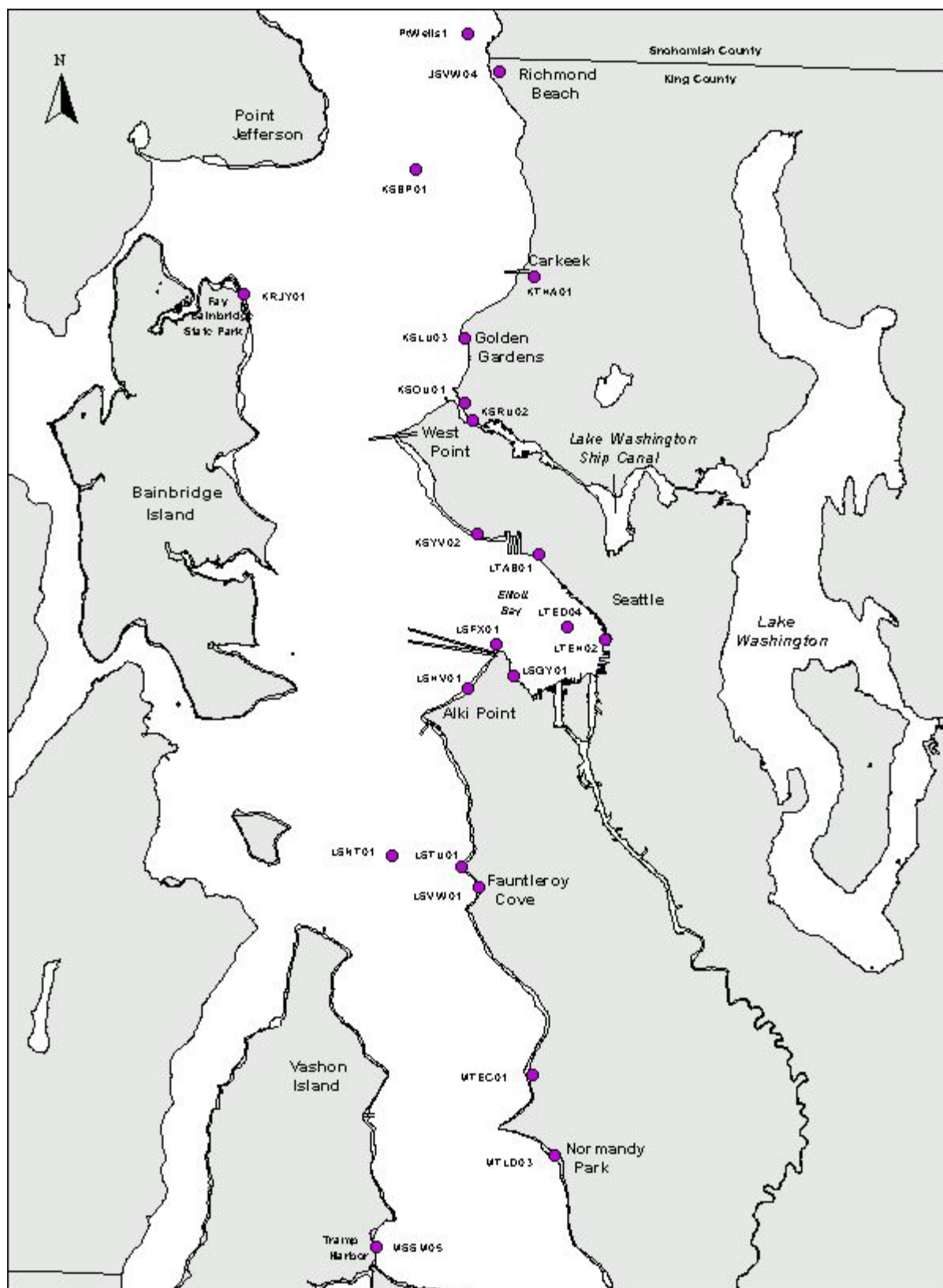
### **2.1.2 Marine Ambient Monitoring Program**

The 2001 ambient monitoring program included collecting water samples at beach, nearshore, and offshore stations and sediments at beach stations (see Table 2-1). Water samples were collected from multiple depths at the offshore and nearshore stations and from a single depth at the beach stations. Shellfish and algae samples were also collected. A total of 15 beach stations and 6 offshore/nearshore stations were sampled in 2001. This program provides background information for comparison of data obtained from the King County point source monitoring program and contributes to a long-term dataset which enables trends to be evaluated regarding overall Puget Sound water quality.

Parameters measured for water samples included physical properties (water clarity, salinity, density, and temperature), and nutrient abundance (nitrogen and phosphorus compounds as well as silica). Dissolved oxygen, chlorophyll, and photosynthetically active radiation were also monitored at some sites. Bacteria was analyzed in all samples where water was collected. Organic compounds (e.g., polynuclear aromatic hydrocarbons, pesticides, and polychlorinated biphenyls), metals, and conventional parameters (total organic carbon, total solids, total volatile solids, and grain size) were monitored in beach sediments. Organic compounds and metals were monitored in shellfish tissues from two stations and only metals were monitored in macroalgae samples. Bacteria was also analyzed in shellfish tissues. Figure 2-2 shows ambient monitoring station locations and Table 2-3 provides a summary of parameters and frequency measured at each station. Station coordinates are provided in Appendix F. Samples at beach sites were collected in August for the annual shellfish, macroalgae, and sediment monitoring, as has been done in past years. At sites where shellfish tissues were analyzed for bacteria five times per year, the samples were collected monthly from May to September. Water samples that were collected for bacteria, physical parameters, and nutrient analyses were analyzed once monthly throughout the year.

## **2.2 MOSS Sampling Program**

The MOSS program is being conducted to assist with siting and design of a new marine outfall. The sampling program includes four major study components: physical oceanography, submarine geophysics, water column profiling, and nearshore habitat and water quality. Data collected for the physical oceanography, submarine geophysics, and nearshore habitat components will be reported in separate documents. Data for water column profiling and beach water quality are provided in this report. Results for water column profiling and beach water quality samples collected in 1999 and 2000 were reported in the *Water Quality Status Report for Marine Waters, 1999 and 2000* (King County, 2001). Offshore and nearshore sediments were



**Figure 2-2. Ambient Monitoring Station Locations**

**Table 2-3. 2001 Ambient and MOSS Stations, Parameters, and Frequency Measured**

(page 1 of 2)

STATION	LOCATION	OFFSHORE/ NEARSHORE/ BEACH	SEDIMENT			WATER				SHELLFISH			ALGAE
			Organics	Metals	Conventional	Bacteria	GWQP *	Organics	Metals **	Organics	Metals	Bacteria	Metals
ITMUKILTSP	Mukilteo Pk.	Beach				◆ 10	◆ 10		◆ 4				
ADMIRALC14	Admiralty Inlet	Offshore				◆ 6	◆ 6						
POSSESSC14	Possession Sound	Offshore				◆ 6	◆ 6						
ITPICNICPT	Picnic Point	Beach				◆ 12	◆ 12		◆ 4				
ITMEADOWBP	Meadowdale Pk	Beach				◆ 12	◆ 12	◆ 2	◆ 6				
EDMDS-CTD4	Edmonds	Offshore				◆ 8	◆ 8						
EDMDS-CTD2	Edmonds	Offshore				◆ 8	◆ 8						
ITOCEANAVE	Ocean Avenue	Beach				◆ 12	◆ 12		◆ 4				
ITBRACKETT	Brackett's Landing	Beach				◆ 12	◆ 12		◆ 4				
ITEDWARDSPT	Edwards point	Beach				◆ 12	◆ 12	◆ 2	◆ 6				
PTWells1	Point Wells	Offshore				◆ 12	◆ 12						
JSVW04	Richmond Beach	Beach	◆ 1	◆ 1	◆ 1	◆ 12	◆ 12		◆ 5	◆ 1	◆ 1	◆ 5	◆ 1
ITBOEINGCR	Boeing Creek area	Beach				◆ 12	◆ 12		◆ 4				
KSBP01	Point Jefferson	Offshore				◆ 12	◆ 12						
ITCARKEEKP	Carkeek Park	Beach				◆ 12	◆ 12	◆ 2	◆ 6				
KTHA01	Piper's Creek	Creek				◆ 12	◆ 12						
KRJY01	Fay Bainbridge	Beach				◆ 6						◆ 5	
KSJX02	Blue Ridge	Beach				◆ 12	◆ 12		◆ 4				
KSLU03	Golden Gardens	Beach				◆ 12	◆ 12		◆ 5	◆ 1	◆ 1	◆ 5	◆ 1
KSQU01	Shilshole Bay	Beach				◆ 12	◆ 12		◆ 4				
KSRU02	LW Ship Canal	Nearshore				◆ 12	◆ 12						
KSYV02	Magnolia	Beach	◆ 1	◆ 1	◆ 1	◆ 12	◆ 12						◆ 1
LTAB01	inner Elliott Bay	Beach				◆ 12	◆ 12						
LTEH02	inner Elliott Bay	Nearshore				◆ 12	◆ 12						
LSGY01	Seacrest	Beach				◆ 12	◆ 12						
LSFX01	Duwamish Head	Beach				◆ 12	◆ 12						
LTED04	Elliott Bay	Offshore				◆ 12	◆ 12						
LSHV01	West Seattle	Beach				◆ 12	◆ 12						



**Table 2-3. 2001 Ambient and MOSS Stations, Parameters, and Frequency Measured**

(page 2 of 2)

STATION	LOCATION	OFFSHORE/ NEARSHORE/ BEACH	SEDIMENT			WATER				SHELLFISH			ALGAE
			Organics	Metals	Conventionals	Bacteria	GWQP *	Organics	Metals **	Organics	Metals	Bacteria	Metals
LSNT01	Dolphin Point	Offshore				◆ 12	◆ 12						
LSTU01	Lincoln Park	Beach				◆ 12	◆ 12						
LSVW01	Fauntleroy Cove	Beach				◆ 12	◆ 12						
COLVOSPASS	Colvos Passage	Offshore				◆ 1	◆ 1						
MTEC01	Seahurst Park	Beach				◆ 6						◆ 5	
MTLD03	Normandy Park	Beach	◆ 1	◆ 1	◆ 1	◆ 6			◆ 1	◆ 1	◆ 1	◆ 5	◆ 1
MSSM05	Tramp Harbor	Beach				◆ 6						◆ 5	

\* GWQP = general water quality parameters. Includes nutrients, salinity, temperature, chlorophyll, dissolved oxygen, solids, transparency, photosynthetically active radiation for offshore/nearshore waters. Includes nutrients, salinity, temperature for beach waters. MOSS beach stations also include pH and dissolved oxygen.

\*\* Total and dissolved metals with the exception of MTLD03 which was dissolved metals only.

MOSS stations

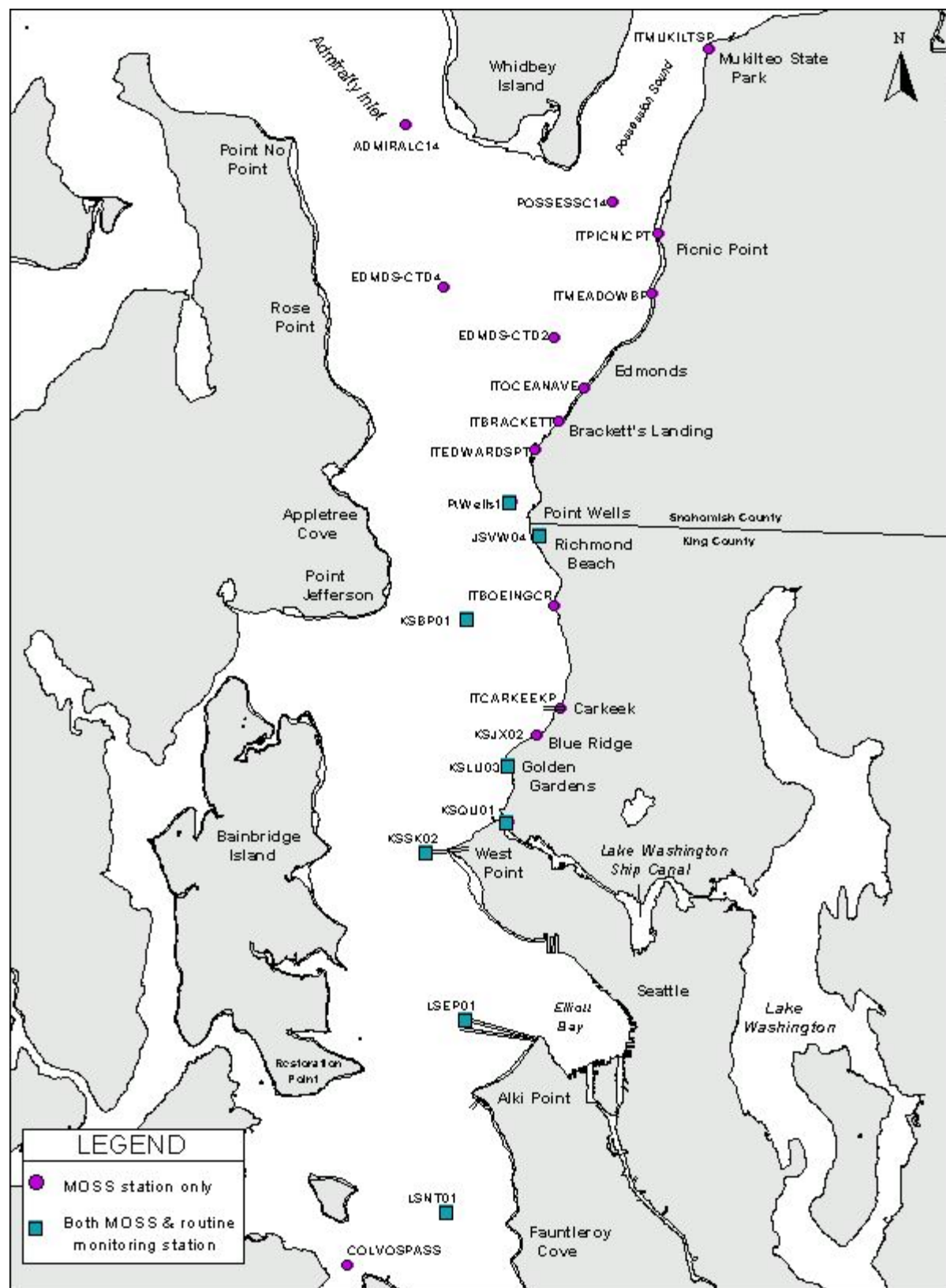
Ambient as well as MOSS program stations

collected in late 2001 and data results are provided in a separate report (King County, 2002). Water column profiling and beach water quality results for 2001 are provided in this report. MOSS station locations are shown in Figure 2-3.

## 2.2.1 MOSS Water Column

The water column profiling component was initiated in 1999 to evaluate physical, chemical, and bacterial characteristics of ambient receiving water in the Central Basin from southern Whidbey Island (including portions of Admiralty Inlet and Possession Sound) to northern Vashon Island. A total of 10 offshore stations (including 3 also sampled for the ambient program and 2 for the point source program) were sampled for physical properties (water clarity, salinity, density, and temperature), and nutrient abundance (nitrogen and phosphorus compounds as well as silica) (see Figure 2-3). Dissolved oxygen, chlorophyll, photosynthetically active radiation, and bacteria (fecal coliform, enterococcus, and *E. coli*) were also monitored.

In addition to the data collection listed above, *in situ* water quality data were collected in a series of transects at five locations in the study area (figure showing transect locations is provided in Section 4). These data were collected to provide information on the physical water quality



**Figure 2-3. Moss Station Locations**

properties of the central Puget Sound basin needed for modeling efforts. Each transect consisted of either five or six stations spaced equidistant across the length of the transect. Measurements were recorded monthly from the surface to a depth of approximately five meters above the bottom. Data collected included salinity, temperature, density, dissolved oxygen, turbidity, and fluorescence. Data collection began in 1999 for the Possession Sound and Point Wells transects and in 2000 for the West Point, Alki, and Edmonds transects. Transect sampling ended in July 2001.

### **2.2.2 MOSS Beach Water Quality**

The beach monitoring component was initiated in March 2000 to evaluate physical, chemical, and bacterial characteristics of nearshore waters in the study area (see Figure 2-3). The station locations were selected to provide coverage of the entire MOSS study area and to assess water quality conditions in areas of particular interest for both ecological and human health risk assessments. Twelve beach stations were sampled in 2001 for the MOSS program, three which were also sampled as part of the ambient sampling program. Nine sites extending from Mukilteo State Park to Blue Ridge were sampled starting in 2000, with the exception of the Mukilteo State Park station which was added to the program beginning in March 2001. Beach waters were analyzed for physical properties (salinity and temperature), nutrient abundance (nitrogen, phosphorus compounds, and silica), pH, dissolved oxygen, and bacteria (fecal coliform, enterococcus, and *E. coli*). Dissolved oxygen and pH were parameters that were added to the program in January 2001. Total and dissolved metals were also analyzed at all 12 beach sites quarterly. Three sites, ITMEADOWSP, ITEDWARDSPT, and ITCARKEEKP, were sampled an additional two times for total and dissolved metals. These same three sites were also sampled twice for organic compounds (including chlorinated pesticides and herbicides, organophosphorus pesticides, polychlorinated biphenyls, and polynuclear aromatic hydrocarbons). As stated above, 3 of the 12 stations are also part of the marine ambient monitoring program (JSVW04, KSLU03, and KSQU01). Additional parameters, such as metals, *E. coli*, pH, and dissolved oxygen were included for the MOSS program that are not part of the routine sampling program. Bacteria samples from one station, KSJX02, were previously collected for the marine ambient monitoring program, but data collection from this station was discontinued in 1996. The beach sampling component ended in February 2002.

## **2.3 Water Column Monitoring**

Water column monitoring at ambient and point source sites is an important component of the County's water quality monitoring programs and is structured to detect natural seasonal changes in the water column as well as identify changes from anthropogenic input. General water quality parameters including temperature, salinity, transparency, dissolved oxygen, chlorophyll-*a*, pheopigment, photosynthetically active radiation, ammonia, nitrate+nitrite, total phosphorus, silica, and total suspended solids, are monitored at multiple depths at multiple sites. Temperature, salinity, and bacteria are the only parameters measured at station KSRU02 located near the mouth of the Lake Washington Ship Canal. Bacteria are monitored at all water column monitoring sites.

### 2.3.1 Bacteria

Biologists and agencies responsible for protecting public health define water quality in terms of several variables, including the presence of bacteria. Fecal coliform bacteria are found in the feces of humans and other warm-blooded animals. These bacteria may enter the aquatic environment directly from humans and animals, from agricultural and storm runoff, and from wastewater. Although fecal coliform bacteria are usually not pathogenic, they may occur along with disease-causing bacteria so their presence indicates the potential for pathogens to be present. Generally, a high fecal coliform count suggests that there is a greater possibility for pathogens to be present. Fecal coliform bacteria are typically found in higher numbers than other pathogens and are easier and safer to test in the laboratory.

Regulatory standards have been established for acceptable levels of fecal coliform bacteria for various water uses, including recreation and fish and wildlife habitat. It should be noted that although fecal coliform bacteria are commonly used as an indicator for the presence of pathogens, there are limitations to the use of these data. There is no recognized numeric association between the number of fecal coliform bacteria and the number of pathogens measured in a sample. In addition, as the presence of viruses and naturally occurring toxic organisms (such as certain dinoflagellates) are not indicated by the presence of fecal coliforms, these organisms must be measured independently.

The Washington State Department of Ecology is currently considering proposing changes for marine surface water bacteriological criteria. The proposed changes incorporate the use of enterococci bacteria as an indicator of bacterial contamination in marine water along with the continued use of fecal coliform bacteria for shellfish protection standards. Enterococci bacteria are a subgroup of the fecal streptococci bacteria and like fecal coliform bacteria, are also found in the intestinal tract of warm-blooded mammals and birds. Specific regulatory changes for marine waters may include adopting an enterococcus geometric mean criterion of 35 colony forming units (cfu)/100 ml with no more than ten percent of the samples exceeding 104 cfu/100 ml. However, marine waters meeting the current geometric mean fecal coliform standard of 14 cfu/100 ml would be considered to be in compliance with the proposed enterococcus geometric mean standard. The state is proposing to keep the current fecal coliform bacterial standards to protect commercial and recreational shellfish harvesting.

Another species of fecal coliform bacteria found in the intestinal tract of humans and warm-blooded animals, *Escherichia coli* (*E. coli*), has also been used as an indicator for the presence of pathogens. Although *E. coli* is not routinely monitored for the County's marine monitoring program, it is being measured for the MOSS program and the results will be used to assist in the project's human-health risk assessment. Pathogenic strains of *E. coli* are rarely found in high numbers and the aquatic environment does not generally support their growth.

### 2.3.2 Temperature and Salinity

Water temperature is an important factor in an estuary. As water temperature rises, biological and chemical activity increases while the capacity of water to hold dissolved oxygen decreases.

Water temperature is dependent upon various factors, including depth, season, amount of tidal mixing, wind, storms, amount of freshwater input, and degree of stratification.

Both temperature and salinity influence water column stratification, although salinity is more important in determining stratification in estuaries. Estuaries usually exhibit changes in salinity as freshwater input increases or decreases. Salinity also fluctuates with tides, amount of input of high salinity water from deep Pacific oceanic water, amount of precipitation, and degree of water column mixing from winds. Generally, salinity increases with water depth unless the estuary is well-mixed.

### **2.3.3 Dissolved Oxygen**

Dissolved oxygen is an important factor controlling the presence or absence of marine species. Aquatic plants and animals require a certain amount of oxygen dissolved in the water for respiration and basic metabolic processes. Waters that contain high amounts of dissolved oxygen are generally considered healthy ecosystems and are capable of sustaining various species of aquatic organisms.

Several factors influence dissolved oxygen concentrations. Seasonal climatic fluctuations can cause water temperature to rise in the spring and summer, reducing the capacity of water to hold dissolved oxygen. In winter, deep oceanic water from the Pacific Ocean containing naturally low levels of oxygen enters Puget Sound. Moreover, anthropogenic input of organic matter and phytoplankton decay may decrease levels of oxygen. Bacteria that utilize organic matter for food consume dissolved oxygen. Hypoxia results when the rate of oxygen consumption, mostly by bacteria decomposing organic material in the water column, exceeds the rate of oxygen production by photosynthesis and by replenishment at the air/water interface from the atmosphere. When the system is overloaded with organic material, oxygen consumption by bacteria may increase to the point where conditions can no longer support marine life (eutrophication), putting fish and other aquatic organisms at risk.

### **2.3.4 Transparency**

Transparency, or water clarity, is measured to determine the depth to which enough light penetrates to support plant growth (euphotic zone). Several factors affect transparency including the amount of suspended silt and soil particles and the amount of phytoplankton and zooplankton in the water column. In addition to transparency, total suspended solids are also monitored. Freshwater input (particularly after storms) and wave action also affect transparency. Low transparency conditions which persist over an extended period of time can degrade the health of a water body as the decreased amount of light penetration reduces the area for aquatic plants and primary producers to grow. In addition, many marine organisms feed by filtering water and large amounts of suspended matter may obstruct their filter-feeding systems.

### 2.3.5 Photosynthetically Active Radiation (PAR)

Sunlight consists of a wide spectrum of wavelengths of which only a small proportion can be used for photosynthesis. This small range of light energy available for photosynthesis is in the 400 to 700 nanometer range. Photosynthetically active radiation (also referred to as light intensity) is measured at various depths throughout the water column to determine the amount of light energy available to phytoplankton, macrophytes, and some diatoms for photosynthesis. PAR is an important factor as phytoplankton and other plants can only inhabit those regions in the water column where enough light penetrates to support photosynthesis. Turbidity, waves, and time of year are factors which may affect PAR measurements.

### 2.3.6 Nutrients

The addition of nutrients, such as nitrogen and phosphorus, into marine waters can have a considerable effect on the water quality, particularly for nearshore habitats where nutrient input typically occurs and tends to be confined. Nutrients may enter marine waters from wastewater discharges, nonpoint runoff, and from riverine and oceanic sources. The greatest impact these nutrients may have is the sudden increase in aquatic plant growth.

The amount of light that penetrates the water column and the amount of nutrients in the water column affect phytoplankton growth. Nitrogen is the primary limiting nutrient that determines the growth of phytoplankton in marine waters (Valiela, 1984). Although nitrogen occurs naturally in the marine environment, abnormal increases from sources such as wastewater or fertilizers can cause increases in phytoplankton growth. An increase in phytoplankton biomass may cause a decline in dissolved oxygen as the phytoplankton cells respire and decay. This depression in dissolved oxygen levels can become critical in areas of reduced circulation. The marine waters within King County have not experienced any significant eutrophication problems, mainly due to the high degree of mixing in the Central Basin of Puget Sound (PSWQAT, 2000).

***Nitrogen Compounds.*** Nitrate, nitrite, and ammonia are forms of inorganic nitrogen used by phytoplankton in the aquatic environment. Nitrates and nitrites are formed through the oxidation of ammonia by nitrifying bacteria. As noted above, nitrogen is usually the limiting nutrient in marine waters. Therefore, an increase in nitrogen compounds could lead to phytoplankton blooms. When blooms occur, water conditions (such as reduced water clarity and dissolved oxygen) may become unfavorable for aquatic organisms. Input of nitrogen compounds may originate from sources such as wastewater from municipal discharges and agricultural runoff.

***Phosphorus.*** Phosphorus is an essential element for aquatic plants and a fundamental element in the metabolic process for both plants and animals. Total phosphorus includes both organic phosphorus and inorganic phosphate. Inorganic phosphates are rapidly taken up by algae and other aquatic plants, although phosphates are usually not the limiting nutrient in marine waters. However, large inputs could cause algal blooms which could lead to unfavorable conditions. Sources of phosphorus potentially entering the marine environment include wastewater from

municipal discharges, industrial wastes, nonpoint agricultural runoff, rivers and streams, and the Pacific Ocean.

**Silica.** Silica is a micronutrient needed by diatoms, radiolarians, some sponges and other siliceous organisms for skeletal growth. Silica can be used as an indicator of plankton blooms, along with chlorophyll-*a*, as silica concentrations will decrease in the photic zone from an increase in phytoplankton uptake. Sediments act as a sink for silica which may be regenerated by various physical and biological processes and reused by organisms on the seafloor and in overlying waters.

### 2.3.7 Chlorophyll and Pheopigments

Chlorophyll-*a* is a green pigment used by algae and green plants during the process of photosynthesis to convert light, carbon dioxide, and water to sugar. Chlorophyll-*a* concentration is the most direct indicator of phytoplankton biomass since all marine planktonic algae contain this photosynthetic pigment. However, chlorophyll-*a* concentrations are not an exact measurement of phytoplankton abundance. The ratio of phytoplankton biomass to chlorophyll varies with species, nutritional status, and environmental conditions. Pheopigments, such as pheophorbide-*a* and pheophytin-*a*, are degradation products of chlorophyll and are produced when phytoplankton cells are grazed upon by zooplankton. High concentrations of pheopigments relative to chlorophyll-*a* indicate a high level of grazing in an aquatic ecosystem. Several factors influence phytoplankton abundance, including amount of solar radiation, extent of grazing, water temperature, nutrient availability, and water column stratification.

### 2.3.8 pH

The pH scale is a unit of measure which describes the degree of acidity or alkalinity of a solution by measuring the amount of hydrogen ions. The letters pH stand for 'the power of hydrogen', whose chemical symbol is "H." It is measured on a scale of 0 to 14 with acids in a range from 0 up to 7, with the lower numbers being a stronger acid. Bases are in the range from 7 to 14, with the higher numbers being a stronger base. Seven is considered to be neutral--neither an acid nor a base and is the pH of distilled water. For water quality, pH is an important measurement as the toxicity of various chemicals may be affected by pH. For the MOSS project, pH was measured as it was necessary to have values for this parameter, along with salinity and temperature, in order to compare total ammonia data to the total ammonia criterion, which is pH, temperature, and salinity based. For offshore waters, a pH of 8.0, temperature of 15 °C, and salinity of 30 PSS are used to compare total ammonia data. The nearshore environment is more variable due to freshwater input and it is necessary to have discrete data to accurately compare ammonia data to applicable standards.

### 2.3.9 Water Column Sampling Methods

**Field Methods.** Offshore water column samples were collected in accordance with the *Recommended Guidelines for Sampling Marine Sediment, Water Column, and Tissues in Puget Sound* (PSEP, 1996) by the King County Environmental Services Section. A brief description is provided below.

Offshore water samples were collected from the *R/V Liberty*, a 42-foot (ft) boat, equipped with a hydraulic crane on the rear deck. Offshore water column profiles were sampled using a SeaBird Electronics SBE 25 SEALOGGER conductivity-temperature-depth (CTD) profiler. Parameters measured by the CTD included: temperature, salinity, density, dissolved oxygen, photosynthetically active radiation (PAR), and chlorophyll-*a* (measured by fluorescence). The CTD was lowered into the water using a hydraulic boom and allowed to equilibrate for 5 minutes at the surface before being lowered to a few meters from bottom depth. Five-liter Niskin bottles were mounted onto the CTD for collecting discrete water samples at predetermined depths for nutrients, total suspended solids, and bacteria. The CTD was electronically programmed to trip individual bottles at specific depths. All bottles were programmed to trip as the CTD ascended through the water column. The CTD was then brought on deck and discrete water samples were immediately drawn from the Niskin bottles and placed into appropriate sample containers. Dissolved oxygen samples were immediately preserved with powdered MnSO<sub>4</sub> (manganous sulfate) and AIA (alkali iodide azide), then stored in the dark. With the exception of dissolved oxygen bottles, sample containers were stored on ice until delivered to the King County Environmental Laboratory.

Transparency (water clarity) measurements were collected using an 12-inch diameter black-on-white Secchi disk. Secchi depths were recorded to the nearest 0.1 meter. As readings may vary depending upon environmental conditions (e.g., waves and glare) and the individual collecting the reading, all field crew were trained to collect measurements using the same procedure.

Intertidal (beach) water samples were collected by inverting sampling bottles just above the water surface, then sinking the bottle down to approximately 12-inches below the water surface. The bottles are not filled completely in order to allow room for mixing. Samples were collected from approximately knee-deep water when possible. At some sites where accessibility is difficult, such as LTAB01 located in inner Elliott Bay, samples were collected with a container lowered on a rope from a pier and then transferred into the sample container. Trace metal samples were collected using non-metallic equipment and employing the "clean hands/dirty hands" technique in accordance with EPA Method 1669 (EPA, 1995).

**Laboratory Methods.** With the exception of temperature, Secchi disk transparency, and CTD parameters which were measured in the field, all water column parameters were analyzed at the King County Environmental Laboratory. Laboratory methods and detection limits are provided in Table 2-4.

Fecal coliform and enterococcus bacteria were analyzed according to Standard Methods 9222D and 9230C, respectively (APHA, 1995). *E. coli* was analyzed according to King County Environmental Laboratory Standard Operating Procedure (SOP) 6.5.1.



All samples were analyzed within their respective hold times and quality assurance/quality control procedures included the use of blanks, duplicates, and spikes where appropriate. All data were reviewed by section supervisors prior to entry into the LIMS (Laboratory Information Management System) database.

**Table 2-4.** Laboratory Methods and Detection Limits for Water Samples

Parameter	Units <sup>1</sup>	MDL <sup>2</sup>	RDL <sup>3</sup>	Method
Salinity	pss	0.005	0.01	SM2520-B
Dissolved Oxygen	mg/L	0.5	1.0	SM4500-O-C
Chlorophyll-a	mg/m <sup>3</sup>	0.01	0.02	EPA 445.0
Phaeophytin	mg/m <sup>3</sup>	0.01	0.02	EPA 445.0
Ammonia-Nitrogen	mg/L	0.01	0.02	SM4500-NH3-H
Nitrite+Nitrate (NO <sub>3</sub> +NO <sub>2</sub> )	mg/L	0.02	0.04	SM4500-NO3-F
Total Phosphorous	mg/L	0.005	0.01	SM4500-P-B,E
Total Suspended Solids (TSS)	mg/L	0.5	1.0	SM2540-D
Silica	mg/L	0.05	0.1	SM4500-SI-D
Turbidity	FTU	0.5	1.0	SM2310-B
pH	pH units	--	--	SM4500-H-B
Fecal coliform	cfu/100 ml	--	--	SM9222-D
Enterococcus	cfu/100 ml	--	--	SM9230-C
<i>E. coli</i>	cfu/100 ml	--	--	Metro SOP 6.5.1
Metals, total & dissolved	µg/L	variable <sup>4</sup>	variable <sup>4</sup>	EPA 1640 (ICP-MS <sup>5</sup> )
Mercury	µg/L	0.0001	0.0005	EPA 1631 (CVAFS)
Semi-volatile organics	µg/L	variable <sup>4</sup>	variable <sup>4</sup>	EPA 8270C
Chlorinated pesticides/PCBs	µg/L	variable <sup>4</sup>	variable <sup>4</sup>	EPA 608
Chlorinated herbicides	µg/L	variable <sup>4</sup>	variable <sup>4</sup>	EPA 8151 (mod.)
Organophosphorus pesticides	µg/L	variable <sup>4</sup>	variable <sup>4</sup>	EPA 8141A

<sup>1</sup>pss = practical salinity scale

mg/L = milligram per liter

µg/L = microgram per liter

mg/m<sup>3</sup> = milligram per meter cubed

cfu = colony forming unit

<sup>2</sup> MDL = method detection limit

<sup>3</sup> RDL = reported detection limit

<sup>4</sup> Detection limits vary with parameter analyzed. Detection limits for individual samples and analytes are provided in the Appendix A.

<sup>5</sup> Sample preparation was by reductive precipitation.

## **2.4 Sediment Monitoring**

Sediment monitoring is a component of the County's ambient and point source monitoring programs as many pollutants (organics and trace metals) tend to be associated with particles that settle out onto bottom sediments. At sufficient concentrations, these compounds may be harmful to benthic organisms and may bioaccumulate. Conventional parameters (total solids, total volatile solids, grain size distribution, total organic carbon, and total sulfides) are also monitored as these parameters affect the bioavailability and/or toxicity of pollutants as well as influence the concentration of pollutants accumulated. A more detailed description of why sediment conventional parameters are measured is provided below.

### **2.4.1 Total Solids**

Total solids are the inorganic and organic particles remaining after a sediment sample has been dried. This parameter is measured in order to convert chemical concentrations from a wet weight to a dry weight basis for reporting uniformity.

### **2.4.2 Total Volatile Solids**

Volatile solids are primarily organic solids that burn in the presence of oxygen at a given temperature (usually 550 or 600 °C). The solids or ash remaining behind is comprised of the non-volatile or fixed solids. The volatile solid value is used as an estimate of organic matter in a sample.

### **2.4.3 Grain Size Distribution**

This is a measure of the size range of particles contained in a given sample. Grain size is usually separated into four main categories: silt, clay, sand, and gravel. Grain size has an influence on chemical concentrations found in sediments and those sediments with a large proportion of small particles (silt and clay) tend to have higher chemical concentrations.

### **2.4.4 Total Organic Carbon**

This is a measure of the total amount of particulate and nonparticulate organic carbon contained in a sample. In the same manner as grain size, total organic carbon also has an influence on chemical concentrations contained in sediments. The higher the organic carbon content, the higher some chemical concentrations tend to be. This is particularly true for organic compounds.

## 2.4.5 Total Sulfides

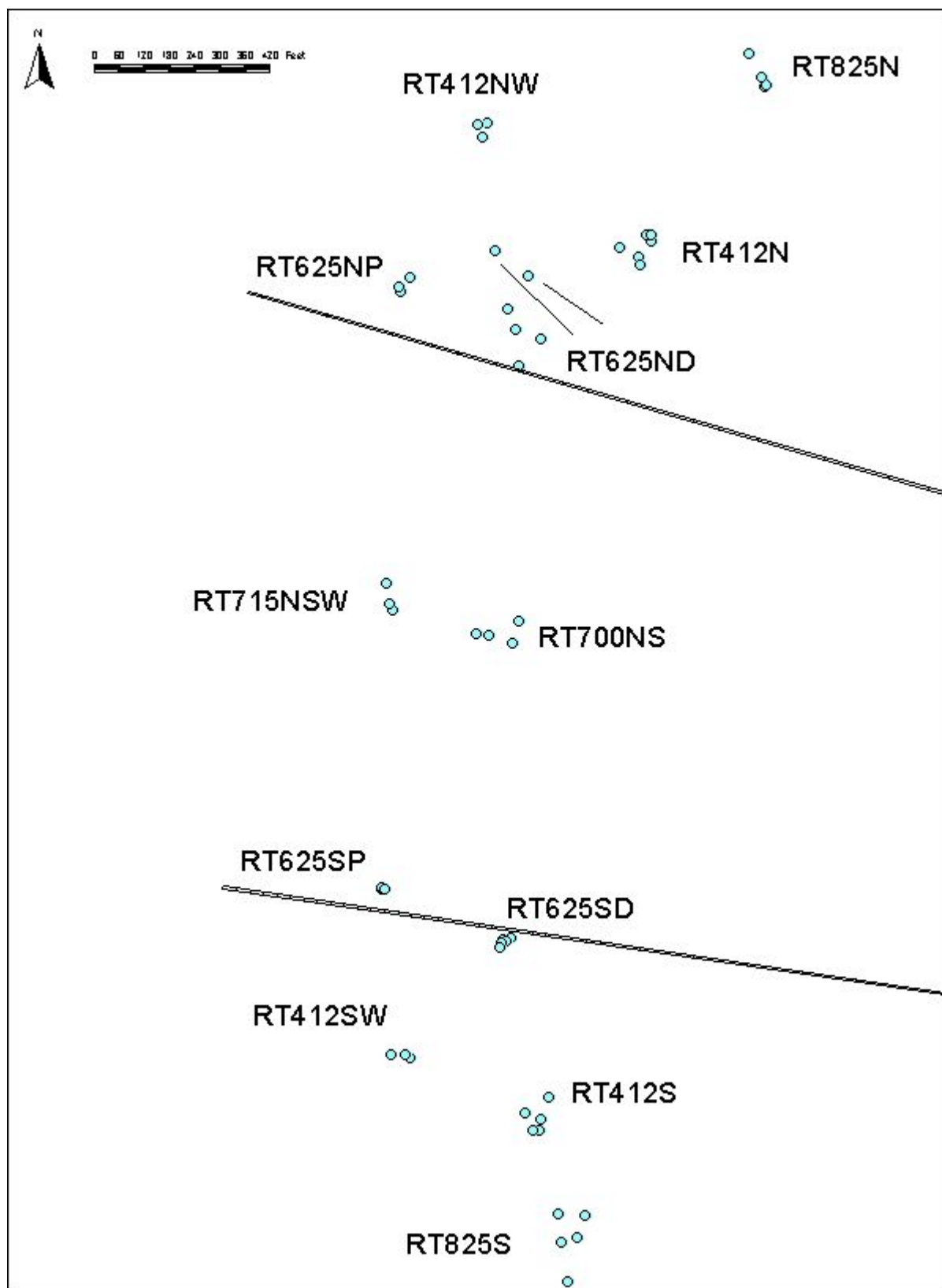
Sulfides are formed by the anaerobic breakdown of organic matter. Total sulfides represent the amount of all sulfide compounds in a given sample and are measured as they may be toxic to some benthic organisms at low concentrations and can create unaesthetic conditions for humans.

## 2.4.6 Sediment Sampling Methods

**Field Methods.** Offshore sediment samples at the Alki CSO TP outfall were collected by the King County Environmental Services Section from the *R/V Liberty*. Samples were collected with two stainless steel 0.1-m<sup>2</sup> modified van Veen grab samplers deployed in tandem. The sampler was decontaminated between sites by scrubbing with a brush to remove excess sediment, followed by an on-board rinsing and thorough *in-situ* rinsing. If sample acceptability criteria were met, the top two centimeters of sediment from a minimum of five subsamples were composited and homogenized before transference to the appropriate sample containers. Sediment samples were collected in accordance with the *Puget Sound Estuary Program (PSEP) Recommended Protocols* (PSEP, 1996) and the County's *Standard Protocol for Marine Sediments* (King County, 1997).

Offshore sediment samples at the South TP outfall were collected as described above with the following exceptions. King County contracted with Blue Water Engineering to monitor the position of the van Veen grab sampler as it contacted the bottom using an underwater acoustic navigation system. King County was interested in obtaining an exact location where the sampler landed on the bottom as opposed to where the boat was positioned above water. Blue Water Engineering used a Trimble Ag132 DGPS for surface navigation, ORE Trackpoint II Plus for acoustic navigation, a motion sensor for boat pitch and roll, and Coastal Oceanographics HYPACK positioning software. The position of the boat was determined using a DGPS and the heading was determined using a digital compass and displayed graphically real-time using a monitor. An acoustic transponder beacon was clamped to the winch wire seven feet above the van Veen sampler and activated prior to deployment. The transponder was clamped high enough above the sampler to avoid interference while sampling. After input of the offsets between the DGPS, the motion sensor and the hydrophone, the HYPACK software was able to calculate the position of the van Veen grab sampler in real-time along with the position of the boat. This enabled the van Veen grab sampler to be 'flown' just above the bottom until the coordinates of the station were reached and then the sampler was lowered. This allowed for accurate sampling of the stations so that the sampler landed 20 feet or less from the stations for most casts for the six stations where this procedure of flying the sampler was attempted. Figure 2-4 shows the exact locations for individual casts at each stations--up to six casts were necessary to obtain enough sediment for all analyses. A detailed description of the sampling methods using this navigational system is provided in *Subsurface Positional Tracking of a Van Veen Sampler in Deep Water Report* (Blue Water Engineering, 2002).

Intertidal (beach) samples were collected by hand-held stainless steel core tubes with a 2-inch diameter. Samples were collected at approximately the same tidal elevation at each beach, 6.5 feet above mean lower low water (MLLW), using a measuring staff, tide chart, and an optical level to sight the proper level on the measuring staff. If the appropriate tidal elevation fell in an



**Figure 2-4.** Location of South Plant Outfall Sediment Grab Samples

area with gravel, cobbles or boulders, then sediments without gravel/cobbles or other large objects closest to the area were sampled instead. Once the required sample amount was obtained, sediments were homogenized in a stainless steel bowl before being transferred to appropriate sample containers. All sampling equipment was pre-cleaned for use at a single site and not reused. All samples were stored on ice until submitted to the laboratory.

**Laboratory Methods.** The King County Environmental Laboratory analyzed all chemical parameters with the exception of particle size distribution and total sulfide. These two analyses were performed by a subcontracted laboratory. Methods and detection limits are provided in Table 2-5. All metals were analyzed using inductively coupled plasma (ICP) emission spectrometry with the exception of mercury. Mercury was analyzed by cold-vapor atomic absorption spectrophotometry (CVAA). Semivolatile organics were extracted with an organic solvent and then analyzed by gas chromatography/mass spectrometry (GC/MS). Pesticides and PCBs were extracted with organic solvents and then analyzed using a gas chromatograph equipped with an electron capture detector (ECD). All samples were analyzed within their respective hold times and quality assurance/quality control procedures included the use of blanks, duplicates, and surrogates and spikes when appropriate. All data were reviewed by section supervisors prior to entry into the LIMS database.

Specific quality assurance/quality control procedures for all offshore sediment analyses can be found in the *2001 NPDES Sediment Baseline Monitoring Plan for the Renton and Alki Outfalls-Sampling and Analysis Plan* (King County, 2001b).

**Table 2-5. Laboratory Methods and Detection Limits for Sediment Parameters**

Parameter	Units	MDL	Method
Total Solids	%	0.005	SM2540-G
Total Volatile Solids	%	0.005	SM2540-G
Total Oil & Grease	mg/kg	100	SM5520-B
Total Organic Carbon	mg/kg	5	SM5310-G
Total Sulfide	mg/kg	10	PSEP, 1986
Metals, total, ICP	mg/kg	variable <sup>1</sup>	EPA 3050/6010
Mercury, total, CVAA	mg/kg	0.019	EPA 7471
Semivolatile Organics	µg/kg	variable <sup>1</sup>	SW 846 8270
Pesticides/PCBs	µg/kg	variable <sup>1</sup>	SW 846 8081/8082
Organotins	µg/kg	0.3	Krone, 1988
Grain Size Distribution	%	0.1	PSEP, 1991

<sup>1</sup>Detection limits vary with parameter analyzed. Detection limits for individual samples and analytes are provided in Appendix B.

## 2.5 Benthic Fauna

Marine benthic communities are useful indicators of sediment quality as the organisms spend the majority of their lives in direct contact with sediments. Benthic organisms can accumulate harmful metal and organic pollutants by ingesting contaminated sediment, eating contaminated prey, or by adsorption from the overlying water, or sediment pore water. There is potential for contaminants to be passed up the food web to other organisms, since benthos are themselves food for fish and other animals.

Benthic communities can be investigated in many ways. Diversity indices, which are a reflection of the numbers and abundances of individuals within a community, are a common and useful method of assessing communities. Total abundance, or the total number of organisms, as well as total species abundance are also useful to describe benthic communities.

Benthic community analysis was conducted in 2001 as part of King County's NPDES monitoring for the South wastewater treatment plant. Samples were collected from six sites around the outfall. Sediment chemistry analysis was conducted concurrent with benthic sampling. Communities were investigated by calculating diversity indices, calculating total and species abundances, looking at populations of pollution tolerant and sensitive species, and by determining the proportion of individuals in major taxonomic groups. Spatial variation and correlation with physical conditions (sediment grain size, water currents) was also explored.

### 2.5.1 Field and Identification Methods

Samples were collected from the entire contents of a single 0.1 m<sup>2</sup> van Veen grab. Three replicates samples were collected at each station. After collection, samples were immediately sieved through a 1 millimeter screen and all material retained on the screen was preserved with 10% buffered formalin. Samples were later rinsed of formalin and stored in 70% alcohol. Preserved animals were sorted into the following major taxonomic groups: Polychaeta, Arthropoda, Mollusca, and miscellaneous (e.g., echinoderms, nemerteans, sipunculans). Samples were then sent to taxonomic experts for further identification.

All animals were identified to the lowest possible taxonomic level, usually species. Identifications were performed by Jeff Cordell (Crustacea), Kathy Welch (Polychaeta), Allan Fukuyama (Mollusca), Valerie Hironaka and Allan Fukuyama (Echinodermata and misc.), and Phil Lambert (Holothuroidea). Quality assurance identifications were performed by Kevin Li (Crustacea), Eugene Ruff (Polychaeta), Susan Weeks (Mollusca), and Scott McEuen (Echinodermata and misc.).

Following identification, the data set was reviewed and some species were eliminated as they represented incidental catches (e.g., nematodes), were pelagic species (e.g., copepods), or were colonial organisms that could not be accurately quantified (e.g., bryozoans and ascidians). It should be noted that colonial organisms were included in the tabulated data as either present or absent but eliminated from the calculation of diversity and abundance indices.

## 2.6 Shellfish and Algae

The uptake of contaminants by marine organisms occurs through ingestion of food and detrital particles, water exchange at feeding and respiratory surfaces, and adsorption of chemicals onto body surfaces. These contaminants may be stored in skeletal material, concretions, and soft tissues (Kennish, 1998). Biological monitoring is a component of the County's ambient and point source monitoring programs, as contaminants may be bioaccumulated by shellfish and algae.

Clam tissues are monitored for organic and metal contaminants and bacteria (fecal coliform and enterococcus). These measurements provide an indication of potential impacts to both shellfish and to humans that consume them. Chlorinated organic compounds (chlorine atoms attached to organic compounds) have been used in pesticides since the 1940s and tend to accumulate in tissue with higher lipid concentrations. Percent lipids in shellfish are also monitored as this parameter affects the concentration of organic pollutants accumulated.

Algae are monitored for metals as it is well documented that algae absorb metals directly from seawater (Phillips, 1994; Hou and Yan, 1998). Algae are used as a biomonitor to assess metal concentrations in intertidal areas.

### 2.6.1 Shellfish and Algae Sampling Methods

**Field Methods.** The King County Environmental Services Section collected shellfish samples. Butter clams (*Saxidomus giganteus*) from each sampling station were collected by hand digging with shovels in the vicinity of siphon holes. A tarp was placed next to the digging site and excavated sediment was placed on the tarp to minimize disturbance to other organisms. The sediment was replaced after clams of sufficient size were removed. After the required number of clams were obtained, they were placed in four-liter glass jars, with the exception noted below, and stored on ice until delivered to the laboratory. A minimum of five butter clams between 60 to 120 millimeters were collected at each station for analyses of metal and organic parameters and composited into a single sample. A minimum of 10 butter clams were collected for the monthly microbiological analyses, then composited into a single sample. Clams that were collected for microbiological analyses were placed in clean, sealable bags rather than glass jars. A minimum of 130 grams of tissue was necessary for analysis.

Algae samples were collected by the King County Environmental Services Section. Algae samples, composed entirely of *Ulva fenestrata*, were collected and placed in 250 ml acid-washed plastic specimen cups and consisted of only attached healthy algae (i.e., discolored or free-floating algae were not collected). The sampling strategy is to collect only the most prevalent edible algae wherever possible, and there was sufficient *Ulva fenestrata* at all the sampling stations to adhere to this strategy. After the required amount of algae were obtained, the containers were stored on ice until delivered to the laboratory.

**Laboratory Methods.** Shellfish samples were processed at the King County Environmental Laboratory in accordance with PSEP recommended protocols (PSEP, 1996). Before the clams

were opened, the shells were rinsed with deionized water to remove sand and other material adhering to the shells. Each clam was measured and the lengths recorded. Tissues from each clam were removed with ceramic blade scalpels, composited with their liquor, and then homogenized with a blender equipped with stainless steel blades. Samples were frozen until analyzed with the exception of the sample portion removed for bacterial analysis. The fecal coliform and enterococcus analyses were initiated immediately following processing.

Algae samples were processed at the King County Environmental Laboratory. Algae were rinsed with deionized water to remove sand and other material adhering to the blades. Sample portions obtained for each station were processed in a blender equipped with titanium blades. Samples were then frozen until analyzed.

The King County Environmental Laboratory analyzed all shellfish and algae parameters. Methods and detection limits are provided in Table 2-6. With the exception of mercury, all metals were analyzed using ICP and/or ICP-MS depending upon detection limit requirements. Mercury was analyzed by cold-vapor atomic absorption spectroscopy. Semi-volatile organics were extracted with an organic solvent and then analyzed by GC/MS. Pesticides and PCBs were extracted with organic solvents and then analyzed using a GC equipped with an ECD. Bacteria samples were processed within eight hours of sample collection and analyzed by multiple-tube fermentation technique.

All samples were analyzed within their respective hold times and quality assurance/quality control procedures included the use of blanks, duplicates, and surrogates and spikes when appropriate. All data were reviewed by section supervisors prior to entry into the LIMS database.

**Table 2-6.** Laboratory Methods and Detection Limits for Shellfish and Algae

Parameter	Units	MDL	Method
Total Solids	%	0.005	SM2540-G
Total Lipids	%	0.1	KCEL OR 07-01-001
Metals, total, ICP	mg/kg	variable <sup>1</sup>	EPA 3050/6010
Mercury, total, CVAA	mg/kg	0.004	EPA 7471
Semivolatile Organics	µg/kg	variable <sup>1</sup>	SW 846 8270
Pesticides/PCBs	µg/kg	variable <sup>1</sup>	SW 846 8081/8082
Fecal Coliform Bacteria	MPN/100g	20	SM9221-E
Enterococcus Bacteria	MPN/100g	20	SM9230-B

<sup>1</sup>Detection limits vary with parameter analyzed. Detection limits for individual samples and analytes are provided in Appendices C and D.



## **2.7 Regulatory Standards**

Regulatory standards and guidelines for water quality have historically focused on those parameters that are of concern to human health. As a result, monitoring programs and criteria were concerned with bacteriological characteristics of surface waters. The focus of water quality guidelines has since expanded to include the health of aquatic organisms, resulting from the widespread use of pesticides, industrial and commercial uses of the Seattle waterfront, and the overall increase in concerns about water quality in Puget Sound. Washington State has implemented wildlife-based water quality standards along with previously existing human health-based standards for surface waters.

Current marine sediment standards are derived from the Apparent Effects Threshold (AET) method (EPA, 1988). This method compares measured chemistry values with associated biological effect data to arrive at empirically-derived chemical concentrations that predict when adverse biological effects should occur. Chemical concentrations below the standard values are predicted to have "no adverse effect". The criteria for marine sediments were developed primarily to protect benthic invertebrates with the assumption that such criteria would also be protective of other species.

The use of bacterial indicators and water quality criteria is necessary in order to evaluate data obtained from monitoring programs. Water quality management decisions can then be based upon these findings. In addition to their use as assessment tools, environmental quality guidelines provide a basis for the development of site-specific water quality objectives for environmental contaminants. These guidelines may also be used to identify the need for source controls to reduce the input of contaminants into marine waters.

The Clean Water Act requires the States to adopt federal water quality criteria or promulgate their own standards which afford equal or better protection to receiving waters.

### **2.7.1 Washington State Standards for Water**

Washington State currently has marine surface water quality standards for conventional pollutants and for some toxics (WAC 173-201A). These standards were derived for the protection of a variety of uses, including human health and the propagation and protection of fish, shellfish, and wildlife. Water quality standards for marine surface waters include, including conventional and toxic pollutants (which includes both acute and chronic values) are provided in Tables 2-7 and 2-8.

### **2.7.2 Washington State Standards for Fecal Coliform Bacteria**

Washington State divides surface water uses into five classes: AA, A, B, C, and Lake. Bacteria concentrations in samples taken from marine waters for both the ambient and point source monitoring programs are compared with the Class AA marine water standard and freshwater samples are compared to the Class AA freshwater standard (Table 2-8).

**Table 2-7. Washington State Marine Surface Water Quality Standards**

Contaminant	Marine Water Quality Standard		Contaminant	Marine Water Quality Standard	
<b>Trace Metals (µg/L)</b>	<b>Acute</b>	<b>Chronic</b>	<b>Semivolatile Organic Compounds (µg/L)</b>	<b>Acute</b>	<b>Chronic</b>
Arsenic <sup>a</sup>	69.0	36.0	Pentachlorophenol	13.0	7.9
Cadmium <sup>a</sup>	42.0	9.3	Total PCBs	10.0	0.030
Chromium VI <sup>a</sup>	1100.0	50.0			
Copper <sup>a</sup>	4.8	3.1			
Lead <sup>a</sup>	210.0	8.1			
Mercury	1.8 <sup>a</sup>	0.025 <sup>b</sup>			
Nickel <sup>a</sup>	74.0	8.2	<b>Other (µg/L)</b>		
Selenium <sup>a</sup>	290	71.0	Ammonia <sup>c</sup> (mg/L)	0.233	0.035
Silver <sup>a</sup>	1.9	---	Chlorine (residual)	13.0	7.5
Zinc <sup>a</sup>	90.0	81.0	Cyanide (weak dissoc.)	1.0	---
<b>Pesticides (µg/L)</b>					
Aldrin/Dieldrin	0.71	0.0019			
Chlordane	0.09	0.004			
Chlorpyrifos	0.011	0.0056			
DDT (and metabolites)	0.13	0.001			
Endosulfan	0.034	0.0087			
Endrin	0.037	0.0023			
Heptachlor	0.053	0.0036			
Lindane	0.16	---			
Toxaphene	0.21	0.0002			
<sup>a</sup> Criteria are based on the dissolved fraction of the metal. <sup>b</sup> Criterion is based on the total recoverable fraction of the metal. <sup>c</sup> Criterion is based on un-ionized ammonia. Source: WAC 173-201a, November 18, 1997.					

The state fecal coliform standards are expressed as geometric mean values. The reason for this is due to the high variability in fecal coliform counts, as bacteria tend to clump and adhere to particulates in water as well as multiply exponentially. Transforming the data using natural logarithms can reduce this variability. This reduces the apparent differences between very high and very low numbers and also simplifies plotting the data by numerically compensating for the exponential growth rate of bacteria. Sample results obtained from King County's monitoring programs are expressed as a moving geometric mean to facilitate comparisons with state bacteria

**Table 2-8. Fecal Coliform Bacteria Standards (colonies/100 ml)**

Class	Moving Geometric Mean <sup>a</sup>	Peak <sup>b</sup>	Comments
AA: Freshwater Marine	50 14	100 43	Exceptional quality suitable for water supply (domestic, industrial, and agricultural), stock watering, fish and shellfish, recreation, and wildlife habitat.
A: Freshwater Marine	100 14	200 43	Can be used for the same purpose as Class AA, but differs in the allowed maximum temperature, minimum level of dissolved oxygen, and pH.
B: Freshwater Marine	200 100	400 200	Listed as "good"; it can be used for industrial and agricultural water supply and secondary contact recreation.
C: Both	200	400	Listed as "fair"; it can be used for industrial water supply, fish migration, secondary contact recreation, commerce, and navigation.
<sup>a</sup> Geometric mean of the 30 most recent samples.			
<sup>b</sup> Not more than 10 percent of the 30 most recent samples may exceed this value.			
Source: WAC 173-201a, 1997; NSSP, 1995.			

standards. This value is obtained by taking the geometric mean value for the 30 most recent samples as directed by the National Shellfish Sanitation Program guidelines for systematic random sampling. As well as the moving geometric mean standard, no more than 10 percent of the samples used to obtain the moving geometric mean value may exceed a defined upper limit. For the Class AA marine water standard this value is 43 colonies/100 ml and 100 colonies/100 ml for the freshwater Class AA standard.

Fecal coliform levels below the method detection limit (MDL) are reported as <MDL. In order to calculate geometric means, any value reported as <MDL was assumed to be one. The moving geometric mean is calculated by taking the results of the 30 most recent samples and applying the formula shown below. When a new value is determined, it becomes part of the moving mean and the oldest value is dropped.

**Computing Geometric Means.**

*Each geometric mean is calculated by taking the sum of the natural logarithms of the sample values, dividing that number by the number of samples, and then taking the inverse natural logarithm. The formula is given below,*

$$\text{geometric mean} = \text{antilog} \frac{1}{n} \sum \log Y$$

*where **n** equals the number of fecal coliform observations and **Y** equals an individual observation (colonies/100 ml).*

### 2.7.3 Washington State Standards for Sediment

Chemicals may occur in sediment as part of the natural environment and from atmospheric deposition, however, sediment may become contaminated by industrial and municipal discharges and non-point sources. Sediment quality guidelines provide a means of assessing sediment quality which leads to informed management decisions regarding sediments and overlying waters.

In 1991, Ecology promulgated the Sediment Management Standards (SMS) which contain numeric criteria for specific organic and metal compounds (Table 2-9). The standards specify, based on the best available knowledge, the levels of sediment contaminants at which no adverse effects to marine organisms are expected. These standards are derived from the Puget Sound Apparent Effects Thresholds (AETs) for selected compounds, which are based on biological testing results (EPA, 1988). Concentrations of compounds that do not exceed the SMS values are not expected to have long-term adverse effects on marine biological resources.

The standards for ionizable organic compounds and metals are presented on a dry weight basis (the wet weight concentration divided by the decimal fraction of the total solids value) while the nonionizable organic compounds are organic carbon normalized (the dry weight concentration in  $\mu\text{g/Kg}$  divided by the dry weight total organic carbon content in  $\text{mg/Kg}$  multiplied by 1000).

The presence of contaminants in sediment does not necessarily indicate that the sediment is toxic to marine organisms. An important factor to the toxicity of contaminants is how much of a toxic compound is available for uptake directly into an organism or accumulated through the food chain. In general, organic compounds, which make up the largest class of chemicals of concern, tend to become associated with the organic matter contained in sediments. The nonpolar, nonionizable organic compounds (such as chlorinated hydrocarbons, aromatic hydrocarbons, and phthalates) have a tendency to adhere to organic matter in water and sediments whereas substances that form ions (such as salts, acids, bases, phenols, and metals) are soluble and therefore dissolve in water.

Organic matter in sediment is a food source for many benthic organisms (organisms that live on or near bottom sediments). Too little organic matter will not support these organisms and too much will reduce the number and/or diversity of organisms due to natural toxic effects associated with enhanced microbial activity. The organic carbon content of sediments has been shown to be related to the bioavailability and toxicity of some organic compounds to aquatic organisms (Di Toro et al., 1991). The toxicity of organic compounds in sediments appears to be more closely correlated to the concentration of organic carbon in the sediments rather than the dry weight concentration. Thus, a more accurate measure of contaminant toxicity can be assessed if the data are “normalized” for the total organic carbon (TOC) content. For this reason, the State standards for nonionizable organics are based upon concentrations that have been TOC normalized (Michelson, 1992). Organic carbon normalization is achieved by dividing the dry weight concentration by the dry weight TOC content. However, when TOC values are very low (e.g.  $<0.2\%$ ) it is not appropriate to normalize contaminant values, as even background levels may exceed regulatory standards. When the TOC content is less than  $0.2\%$ , dry weight values are more appropriate to use than organic carbon normalized values.

**Table 2-9. Washington State Sediment Standards**

Contaminant	Sediment Quality Standard	Lowest Apparent Effects Threshold	Contaminant	Sediment Quality Standard	Lowest Apparent Effects Threshold
<b>Metals</b>	mg/kg dry weight		<b>Nonionizable Organic Compounds</b>	mg/kg organic carbon	µg/kg dry weight
Arsenic	57		1,2-Dichlorobenzene	2.3	35
Cadmium	5.1		1,4-Dichlorobenzene	3.1	110
Chromium	260		1,2,4-Trichlorobenzene	0.81	31
Copper	390		Hexachlorobenzene	0.38	22
Lead	450		Dimethyl phthalate	53	71
Mercury	0.41		Diethyl phthalate	61	200
Silver	6.1		Di-n-butyl phthalate	220	1400
Zinc	410		Butyl benzyl phthalate	4.9	63
			Bis (2-ethylhexyl) phthalate	47	1300
			Di-n-octyl phthalate	58	6200
<b>Nonionizable Organic Compounds</b>	mg/kg organic carbon	µg/kg dry weight	Dibenzofuran	15	540
Total LPAHs <sup>a</sup>	370	5200	Hexachlorobutadiene	3.9	11
Naphthalene	99	2100	N-Nitrosodiphenylamine	11	28
Acenaphthylene	66	1300	Total PCBs	12	130
Acenaphthene	16	500			
Flourene	23	540	<b>Ionizable Organic Compounds</b>	mg/kg dry weight	
Phenanthrene	100	1500	Phenol	0.42	
Anthracene	220	960	2-Methylphenol	0.063	
2-Methylnaphthalene	38	670	4-Methylphenol	0.67	
Total HPAHs <sup>b</sup>	960	12000	2,4-Dimethylphenol	0.029	
Fluoranthene	160	1700	Pentachlorophenol	0.36	
Pyrene	1000	2600	Benzyl alcohol	0.057	
Benzo(a)anthracene	110	1300	Benzoic acid	0.65	
Chrysene	110	1400			
Total Benzofluoranthenes	230	3200			
Benzo(a)pyrene	99	1600			
Indeno(1,2,3-c,d)pyrene	34	600			
Dibenzo(a,h)anthracene	12	230			
Benzo(g,h,i)perylene	31	670			
<sup>a</sup> Represents the sum of the following low molecular weight PAHs: Naphthalene, Acenaphthylene, Acenaphthene, Fluorene, Phenanthrene, and Anthracene. <sup>b</sup> Represents the sum of the following high molecular weight PAHs: Fluoranthene, Pyrene, Chrysene, Benzo(a)anthracene, Benzo(a)pyrene, total Benzofluoranthenes, Indeno(1,2,3-c,d)pyrene, Dibenzo(a,h)anthracene, and Benzo(g,h,i)perylene.					
Source: Ecology, 1995					

## **2.7.4 Standards for Biota**

In addition to contaminants found in water and sediment, several contaminants have the potential to accumulate in the tissues of aquatic biota, such as fish and shellfish. Bioaccumulation in biota may affect not only the species directly accumulating the contaminants, but humans and other species that consume the affected species. Numerical tissue-residue guidelines provide a basis for assessing the hazards that tissue-laden contaminants pose to human health and wildlife, and therefore, a basis for regulating contaminant inputs into the environment. Ecology does not currently have tissue-residue standards, however, heavy metal concentrations in shellfish samples were compared with EPA guidelines.